

APPENDIX E UNITS AND CONVERSION FACTORS

Quantity	Conversion
Length	1 m = 100 cm = 3.28084 ft = 39.3701 in.
Mass	1 kg = 10 ³ g = 2.20462 lb _m
Force	1 N = 1 kg (m/s ²) = 0.224809 lb _f
Pressure	1 bar = 10 ⁵ kg/(m s ²) = 10 ⁵ N/m ² = 10 ⁵ Pa = 10 ² kPa = 0.986923 atm = 14.5038 psia = 7 50.061 torr
Volume	1 m ³ = 10 ⁶ cm ³ = 35.31417 ft ³
Density	1 g/cm ³ = 10 ⁻³ kg/m ³ = 62.4278 lb _m /ft ³
Energy	1 J = 1 kg (m ² /s ²) = 1 N m = 1 m ³ Pa = 10 ⁻⁵ m ³ bar = 10 cm ³ bar = 9.86923 cm ³ (atm) = 0.239006 cal = 5.12197 × 10 ⁻³ ft ³ psia = 0.737562 ft lb _f = 9.47831 × 10 ⁻⁴ (Btu)
Power	1 kW = 10 ³ W = 10 ³ kg (m ² /s ³) = 10 ³ J/s = 239.006 cal/s = 737.562 ft lb _f /s = 0.947831 Btu/s = 1.34102 hp

Values of the universal gas constant *R*

$$\begin{aligned}
 &= 8.314 \text{ J mol}^{-1} \text{ K}^{-1} = 8.314 \text{ m}^3 \text{ Pa mol}^{-1} \text{ K}^{-1} \\
 &= 8.314 \times 10^{-3} \text{ m}^3 \text{ kPa mol}^{-1} \text{ K}^{-1} = 8314 \text{ cm}^3 \text{ kPa mol}^{-1} \text{ K}^{-1} \\
 &= 82.06 \text{ cm}^3 \text{ (atm) mol}^{-1} \text{ K}^{-1} = 62.356 \text{ cm}^3 \text{ (torr) mol}^{-1} \text{ K}^{-1} \\
 &= 1.987(\text{cal}) \text{ mol}^{-1} \text{ K}^{-1} = 1.986(\text{Btu})(\text{lb mole})^{-1} (\text{°R})^{-1} \\
 &= 0.71302 (\text{ft})^3 (\text{atm})(\text{lb mol})^{-1} (\text{°R})^{-1} = 10.73 (\text{ft})^3 (\text{psia}) (\text{lb mol})^{-1} (\text{°R})^{-1} \\
 &= 1,545(\text{ft})(\text{lb}_f)(\text{lb mol})^{-1} (\text{°R})^{-1}
 \end{aligned}$$